

Section 8.a

Modeling Leachate Natural Attenuation

Vejen Landfill, Vejen, Denmark

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Overview

- landfill MNA - a new perspective
- detailed case study
 - Vejen landfill site - leachate natural attenuation
- BioRedox reaction schemes for modeling:
 - Cr(VI) attenuation
 - permeable reactive barriers (PRBs)
 - ORC for bioremediation of BTEX and MTBE

Landfill MNA

- researchers studying leachate natural attenuation since the 1970's
- older municipal landfills
 - typically unlined, uncovered for decades
 - organic solvents discarded with refuse
 - biodegradation in landfill
 - bioattenuation of leachate plumes below LF
 - sometimes natural attenuation has effectively mitigated plume migration for decades

Landfill MNA

- favorable conditions for biostabilization
 - refuse: moisture, high temp., rich in OM
 - aquifer:
 - organic-rich leachate
 - sequential anaerobic-aerobic zones
- remediation of old municipal landfills
 - permeable caps --> bioreactor
 - “biofilter” caps - methanotrophic co-metabolism
 - filter out VOCs in landfill gas
 - destroy pollutants through biodegradation
 - continued bioattenuation of leachate below landfill

Landfill MNA

- remediation of old municipal landfills
 - low-permeability caps --> entombment
 - contain pollutants for hundreds of years
 - technology lifespan uncertain
- movement towards allowing LF MNA in some cases
 - long-term monitoring critical
 - integrated processes between components:
 - cap --> landfill --> aquifer

Landfill MNA

- may need predictive tool to assess long-term effectiveness of landfill MNA
 - integrated remediation alternatives
 - landfill caps
 - landfill gas collection
 - natural attenuation or enhanced bio in groundwater
- reactive transport models
 - insight into effects of integrated technologies

Vejen Landfill Site

- Vejen, Denmark
- unlined landfill over shallow aquifer
- stable leachate plume
- extensive redox zone delineation (t=15 yrs)
- Thomas Christensen et al. (Technical University of Denmark)
 - landfill hydrology
 - hydrogeology, biogeochemistry in aquifer

Vejen Landfill Site

- Monitoring program:
 - 9 borings along a transect (370 m long)
 - 13 to 31 samples collected from each boring (0.5 meter intervals)
 - 10 cm-long iron screens fitted onto 1-inch iron pipes
 - additional borings placed within 130 meters of landfill
 - temporal fluctuations in redox parameters

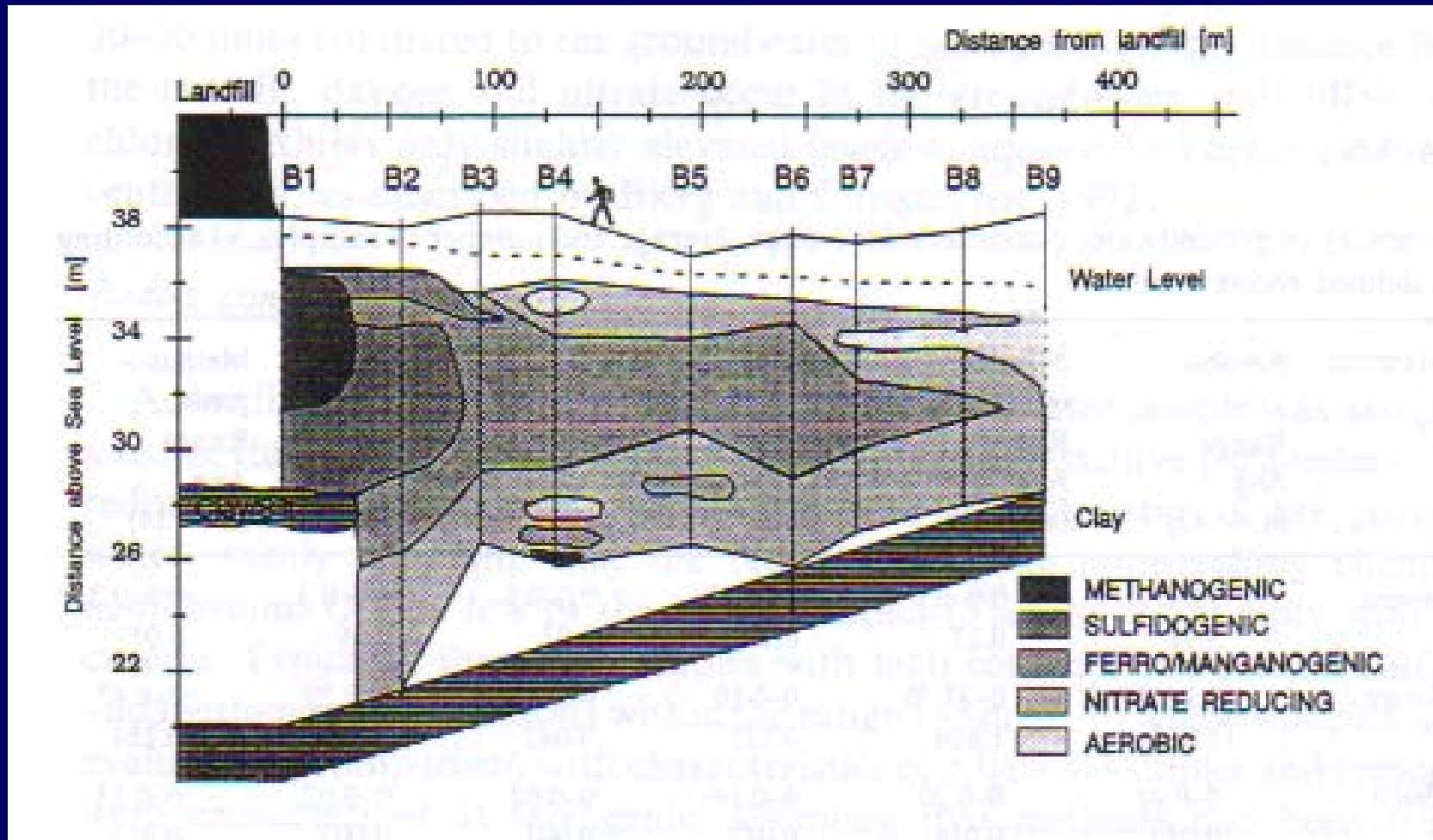
Conceptual Model

- homogeneous shallow aquifer
- leachate source model (TOC)
- groundwater velocity: 500 to 650 ft/year
- 2D cross-section model
 - groundwater flow: MODFLOW
 - solute transport: *BioRedox*

Conceptual Model

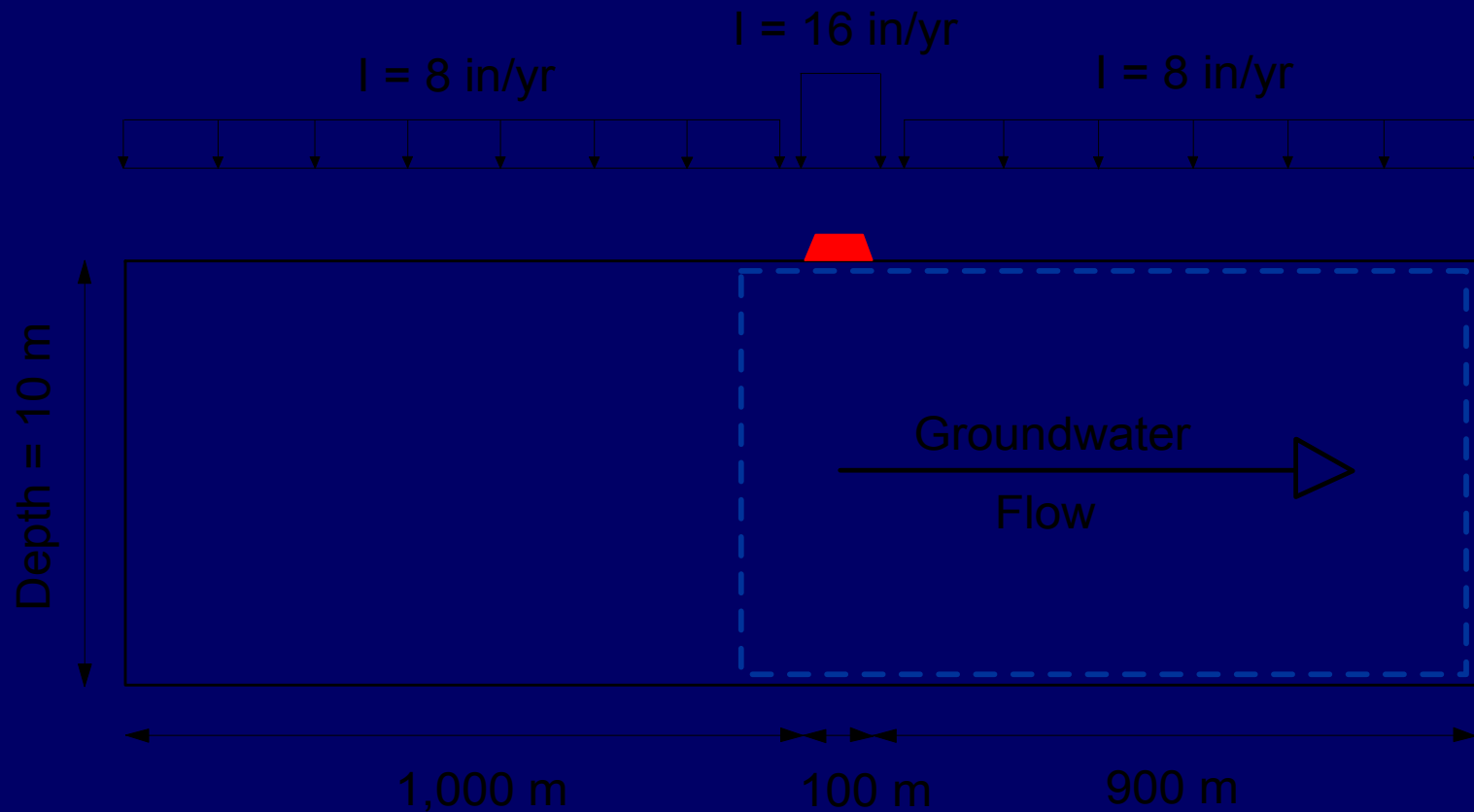
- *BioRedox* represents:
 - leachate source
 - TOC oxidation
 - electron acceptor reduction (O_2 , NO_3 , $Fe(OH)_3$, SO_4 , and CO_2)

Redox Zones in Groundwater

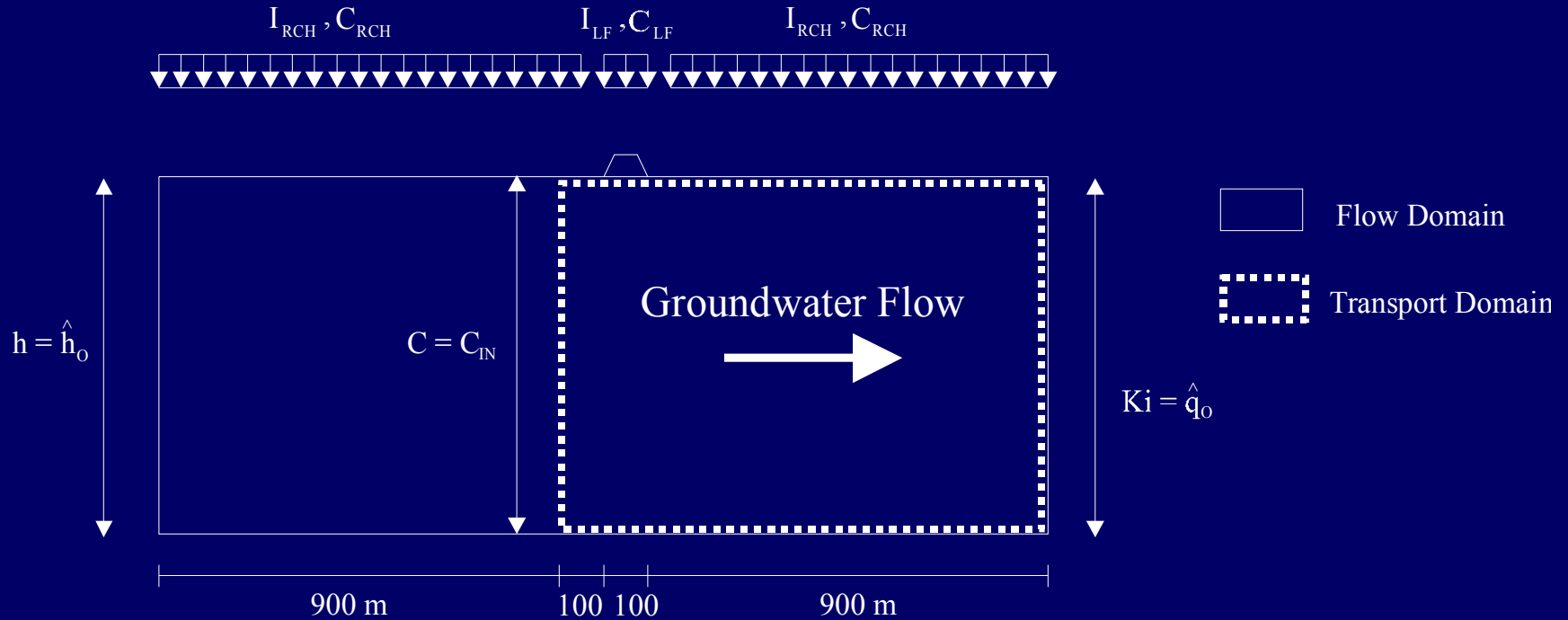


Source: Lyngkilde and Christensen, 1992

Model Domain



Boundary Conditions



Solute Properties

<i>Solute</i>	<i>Molecular Weight</i> (g/mol)	<i>K_{oc}</i> (mL/g)	<i>Units of Concentration</i>	<i>Boundary and Initial Conditions</i>			
				<i>C_{IN}</i>	<i>C_{RCH}</i>	<i>C_{LF}</i>	<i>C(x,t=0)</i>
TOC	12	0	mg/L	0	0	(a)	0
PCE	166	364	ug/L	0	0	250	0
TCE	131.5	126	ug/L	0	0	250	0
DCE	97	49	ug/L	0	0	250	0
Vinyl Chloride	62.5	57	ug/L	0	0	250	0
Oxygen	32	0	mg/L	7	7	0	7
Nitrate	62	0	mg/L	25	25	0	25
Ferric Hydroxide	55.85	N/A	mg/g	N/A	N/A	N/A	5
Sulfate	96.1	0	mg/L	40	40	400	40

(a) $C = 14000e^{-0.26t}$, $t \leq 10$ years; $C = 1040$ mg/L, $t > 10$ years

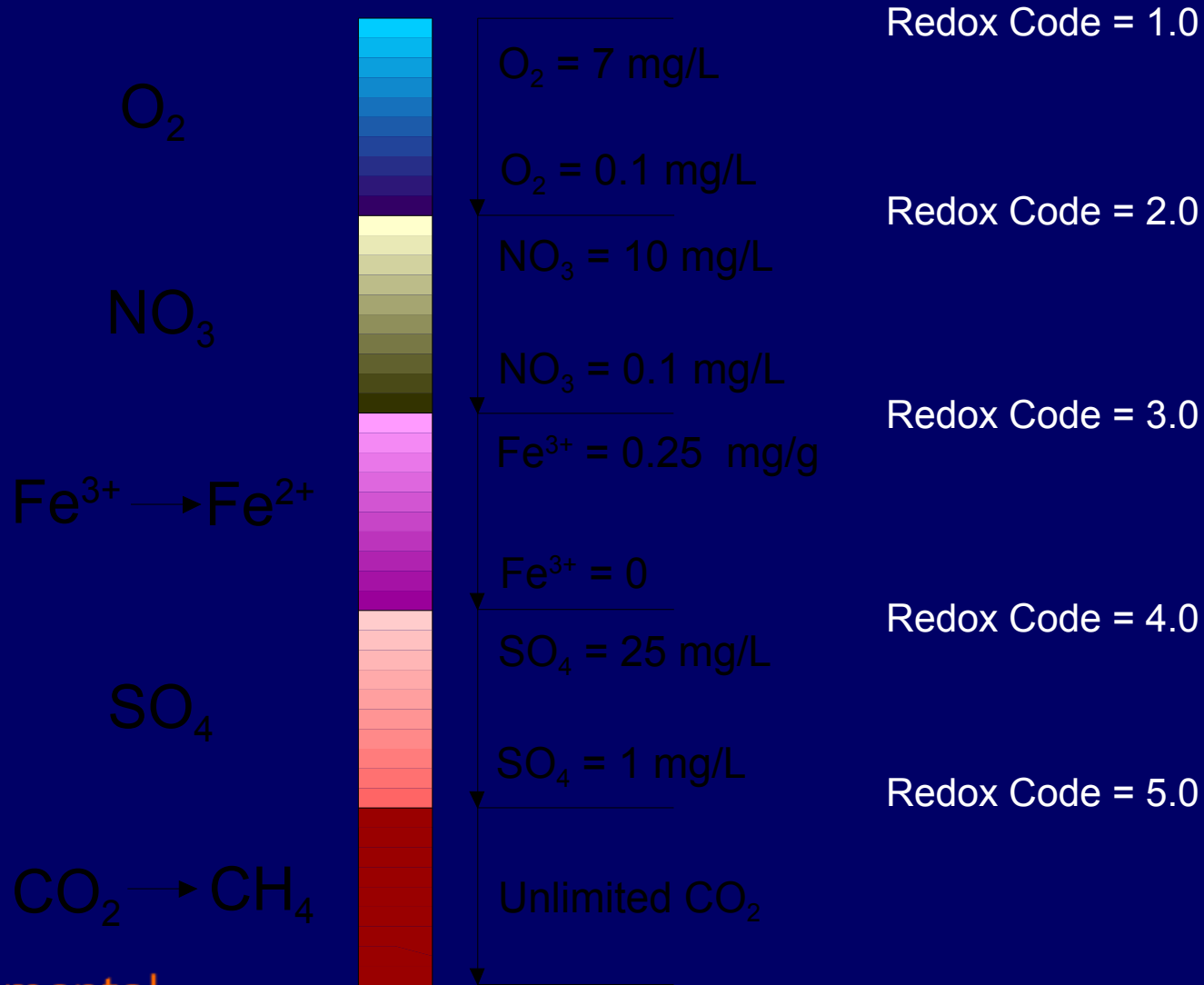
Solute Reactions

<i>Solute</i>	<i>Type</i>	<i>Half-Reaction</i>
TOC	Oxidation	$\text{CH}_2\text{O} + \text{H}_2\text{O} \rightarrow \text{CO}_2 + 4\text{H}^+ + 4\text{e}^-$
Vinyl Chloride	Oxidation	$\text{CH}_2\text{CHCl} + 2\text{H}_2\text{O} \rightarrow \text{CH}_3\text{COOH} + 3\text{H}^+ + 3\text{e}^-$
Oxygen	Reduction	$\text{O}_2 + 4\text{H}^+ + 4\text{e}^- \rightarrow 2\text{H}_2\text{O}$
Nitrate	Reduction	$\text{NO}_3^- + 6\text{H}^+ + 5\text{e}^- \rightarrow 0.5\text{N}_2(\text{g}) + 3\text{H}_2\text{O}$
Ferric Hydroxide	Reduction	$\text{Fe}(\text{OH})_3(\text{s}) + 3\text{H}^+ + \text{e}^- \rightarrow \text{Fe}^{2+} + 3\text{H}_2\text{O}$
Sulfate	Reduction	$\text{SO}_4^{2-} + 9\text{H}^+ + 8\text{e}^- \rightarrow \text{HS}^- + 4\text{H}_2\text{O}$

<i>Organic Solute</i>	<i>Redox Zone Biodegradation Rate</i>				
	<i>CH₄</i>	<i>SO₄</i>	<i>Fe(III)</i>	<i>NO₃</i>	<i>O₂</i>
TOC	k_{TOC}	k_{TOC}	k_{TOC}	k_{TOC}	k_{TOC}
PCE	k_{RD}	$0.75 k_{\text{RD}}$	$0.5 k_{\text{RD}}$	0	0
TCE	k_{RD}	$0.75 k_{\text{RD}}$	$0.5 k_{\text{RD}}$	0	0
DCE	k_{RD}	0	0	0	0
Vinyl Chloride	$0.75 k_{\text{RD}}$	0	k_{OXD}	0	k_{OXD}

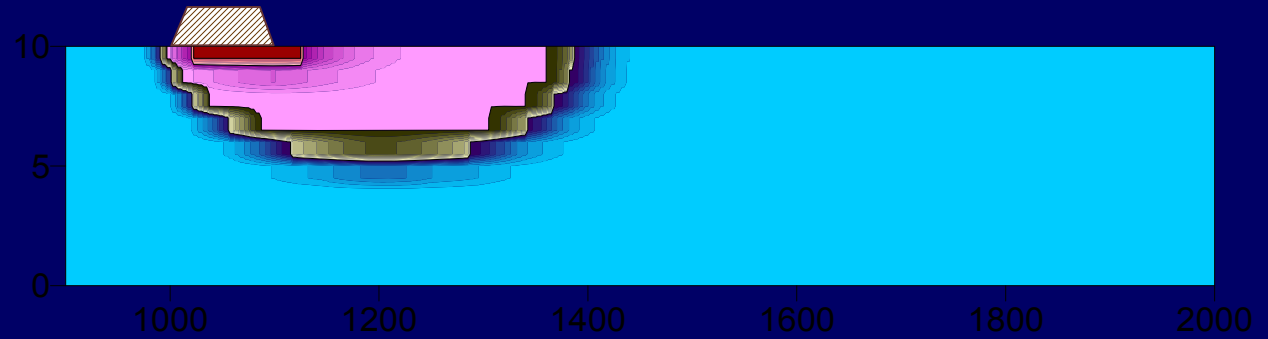
<i>Scenario</i>	<i>k_{TOC}</i> <i>(d⁻¹)</i>	<i>k_{RD}</i> <i>(d⁻¹)</i>	<i>k_{OXD}</i> <i>(d⁻¹)</i>
no cap	7.6E-3	7.6E-3	7.6E-3
high-K cap	7.6E-3	7.6E-3	7.6E-3
low-K cap	7.6E-3	1.9E-3	7.6E-3

Redox Zone Visualization

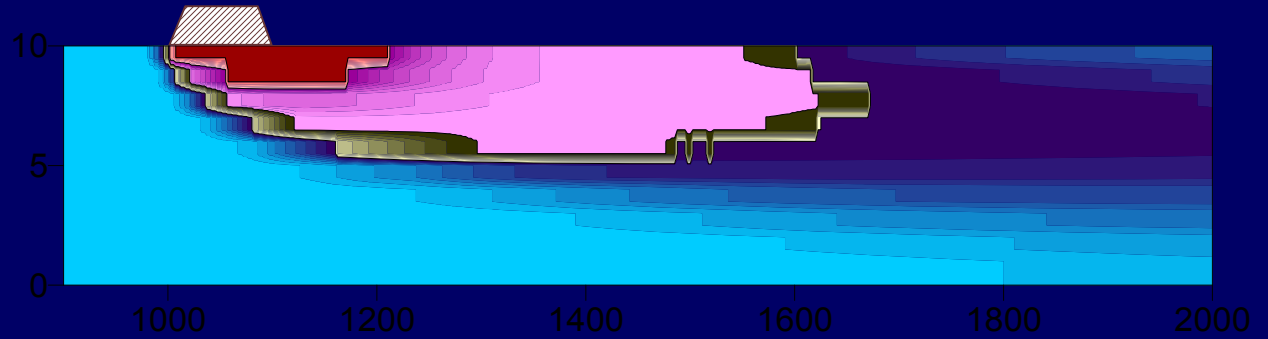


Transient Bioattenuation Capacity

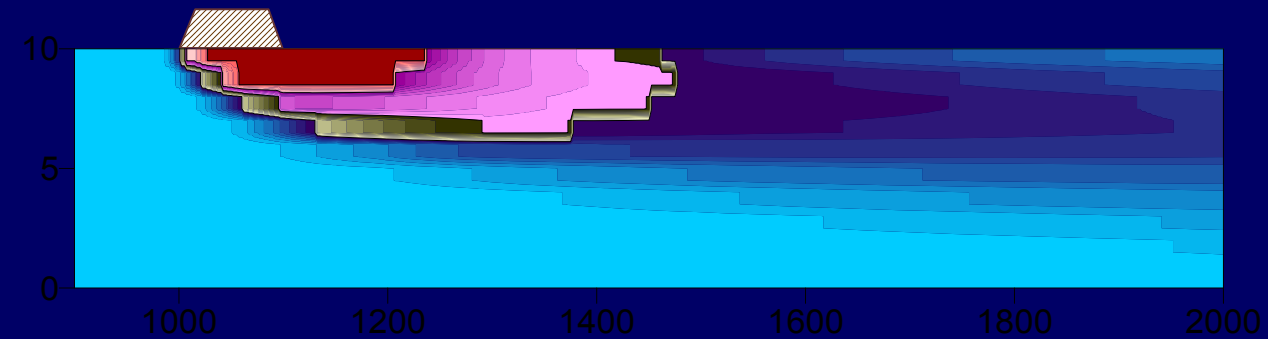
t = 1 year



t = 5 years



t = 15 years



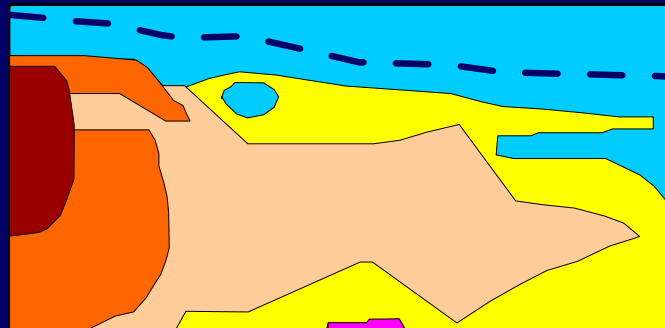
Vejen Landfill Comparison

<i>Redox Zone</i>	<i>Observed</i> *	<i>Simulated</i>
Methanogenic / Sulfate-Reducing	75 metres	100 metres
Ferrogenic / Manganogenic	250 metres	250 metres
Nitrate-Reducing	25 metres	5 metres

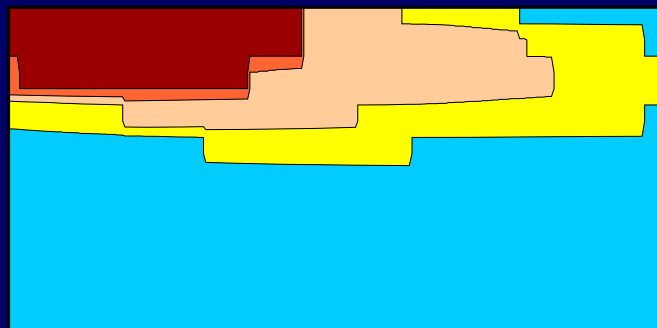
* Lyngkilde and Christensen, 1992, J. Contam. Hydrol., 10: 273-289.

Vertical dispersivity = 0.003 m
rate = 1.0e-2 per day
TOC degrades when TOC > 0

Observed:



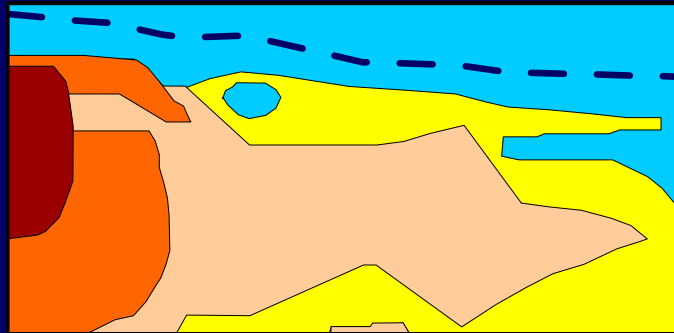
Model:



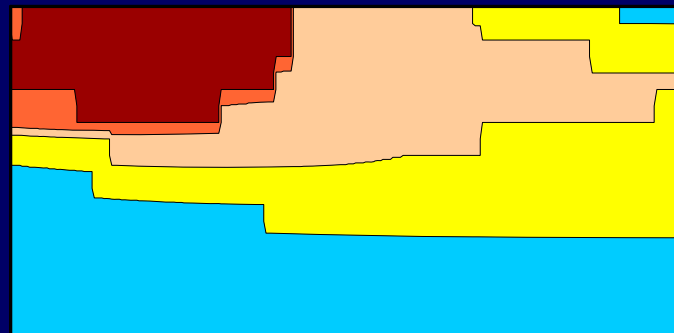
-  aerobic zone
-  nitrate-reducing zone
-  iron(III)-reducing zone
-  sulfate-reducing zone
-  methanogenic zone

Vertical dispersivity = 0.008 m
rate = $7.6e-3$ per day
TOC degrades when TOC > 0.1 mg/L

Observed:








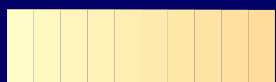
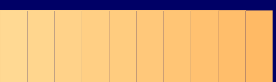

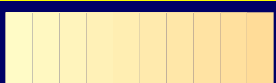
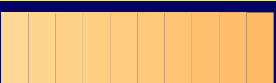
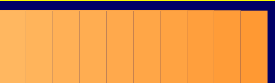
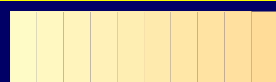
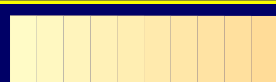
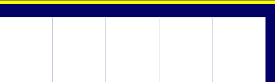
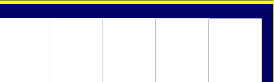
Model:



-  aerobic zone
-  nitrate-reducing zone
-  iron(III)-reducing zone
-  sulfate-reducing zone
-  methanogenic zone

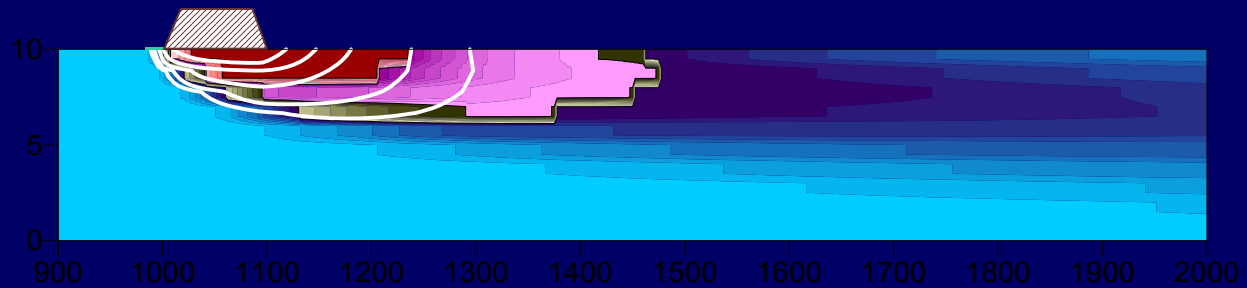
Hypothetical CAH Model

Biodegradation Kinetics

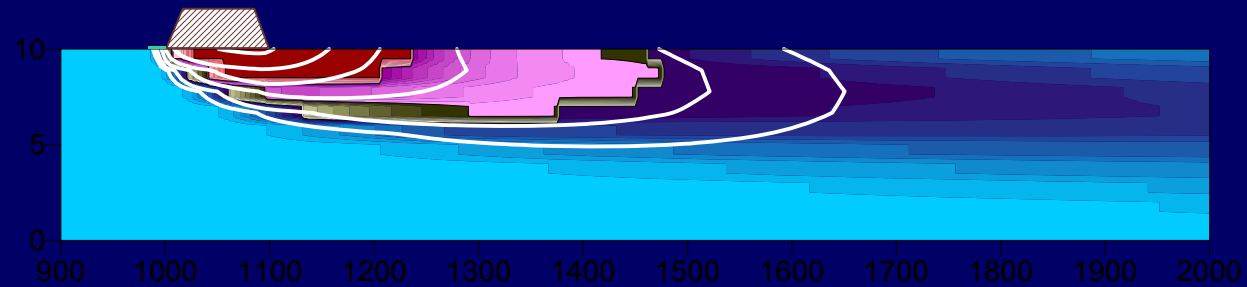
	<i>Redox Zone</i>				
	<i>CH₄</i>	<i>SO₄</i>	<i>Fe(III)</i>	<i>NO₃</i>	<i>O₂</i>
TOC					
PCE					
TCE					
DCE					
VC					

Without Methane Transport:

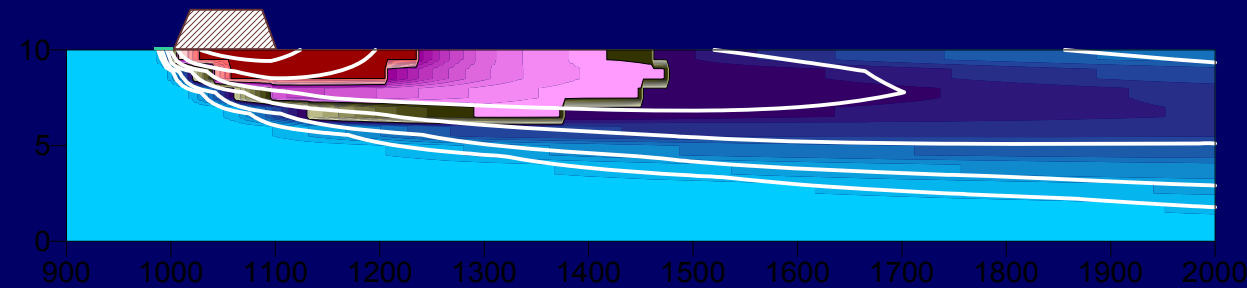
PCE



TCE

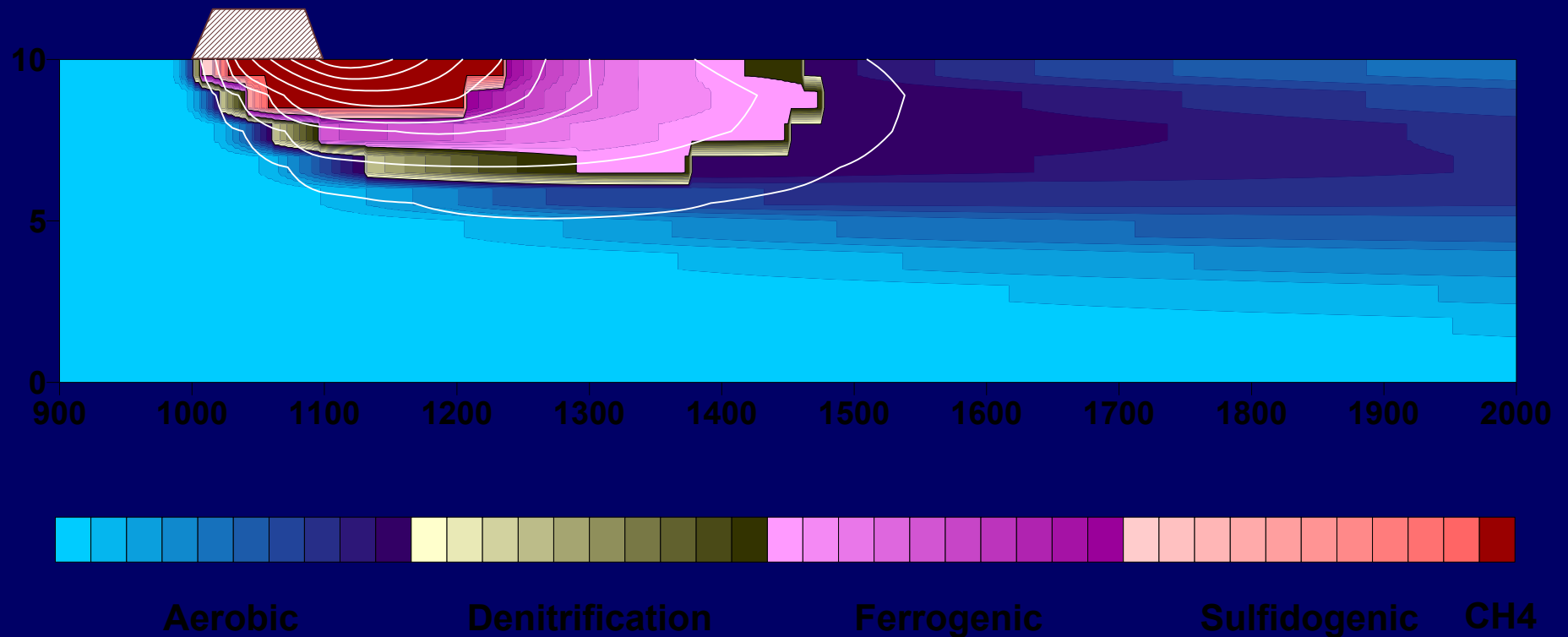


DCE



(CAH concentration contours of 1, 2, 5, 10, 20, 50, and 100 ug/L)

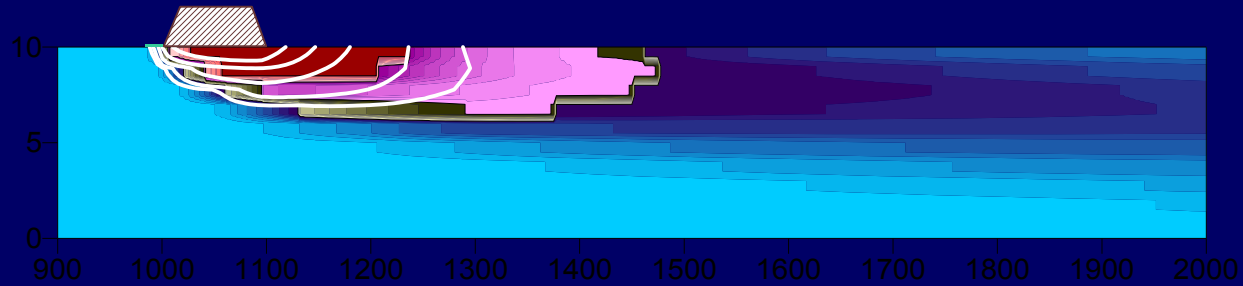
Methane Simulation:



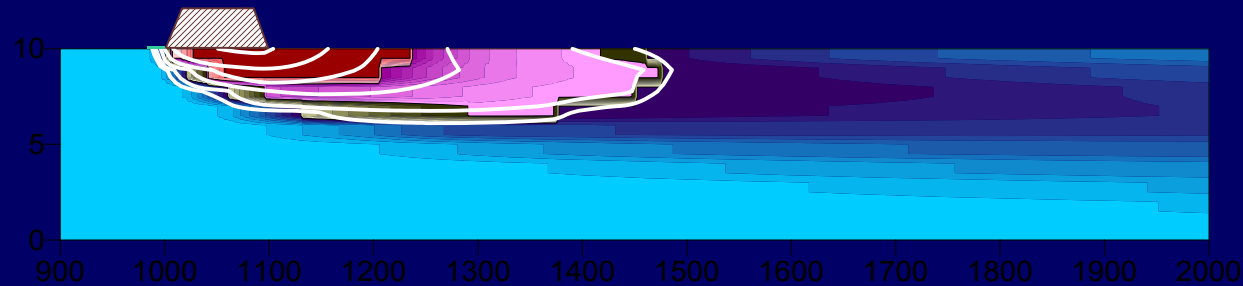
(Methane concentration contours of 0.1, 1, 5, 10, 20, 30, 40, and 50 mg/L)

Methanotrophic Biodegradation:

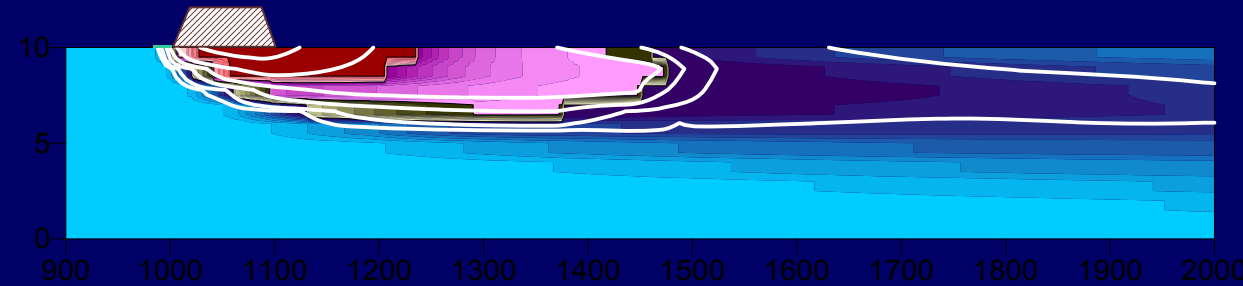
PCE



TCE



DCE



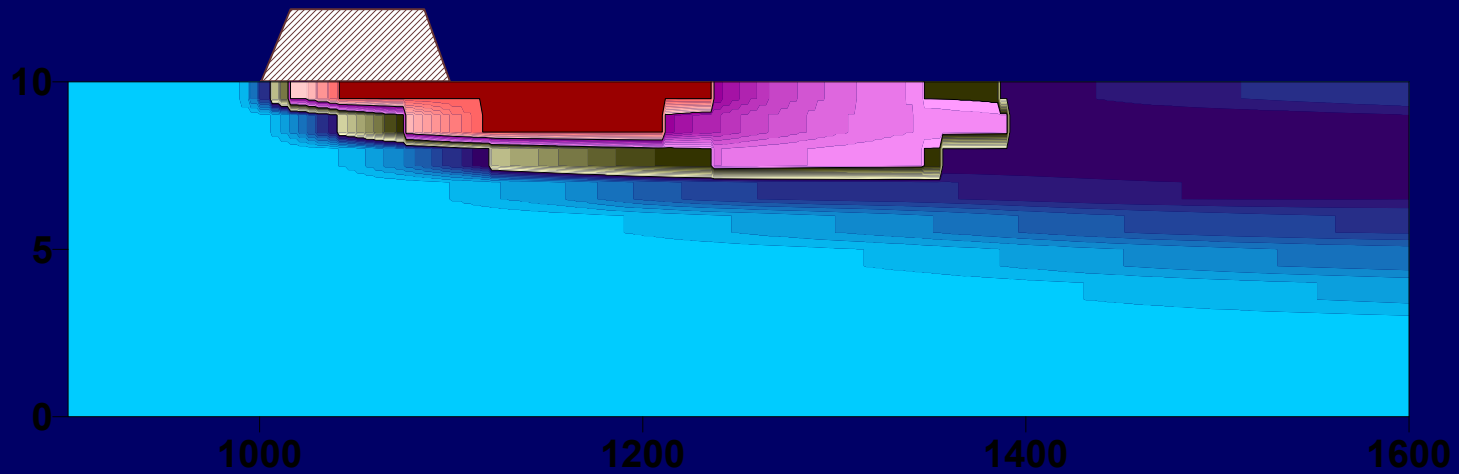
(CAH concentration contours of 1, 2, 5, 10, 20, 50, and 100 ug/L)

Cap Performance Comparison *(t=15 to 30 years)*

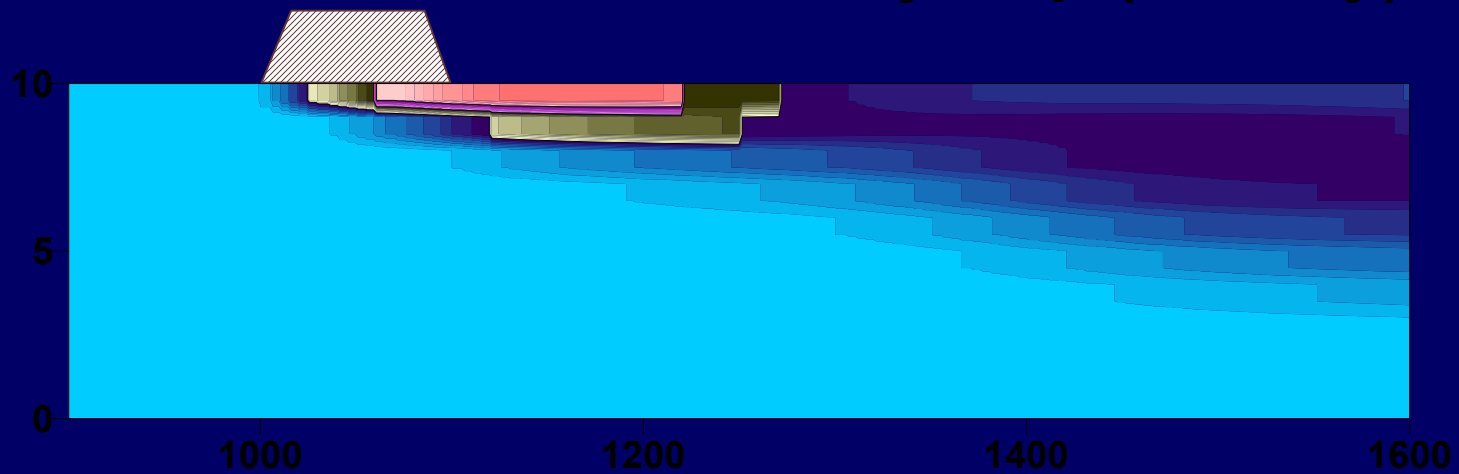
Cap Performance Comparison

- cap placement (t=15 to 30 years)
- two scenarios:
 - Permeable cap: $I = 6 \text{ in/yr}$
 - Clay Cap: $I = 1 \text{ in/yr}$
- compared effects on groundwater impact
 - TCE criteria is 5 ug/L

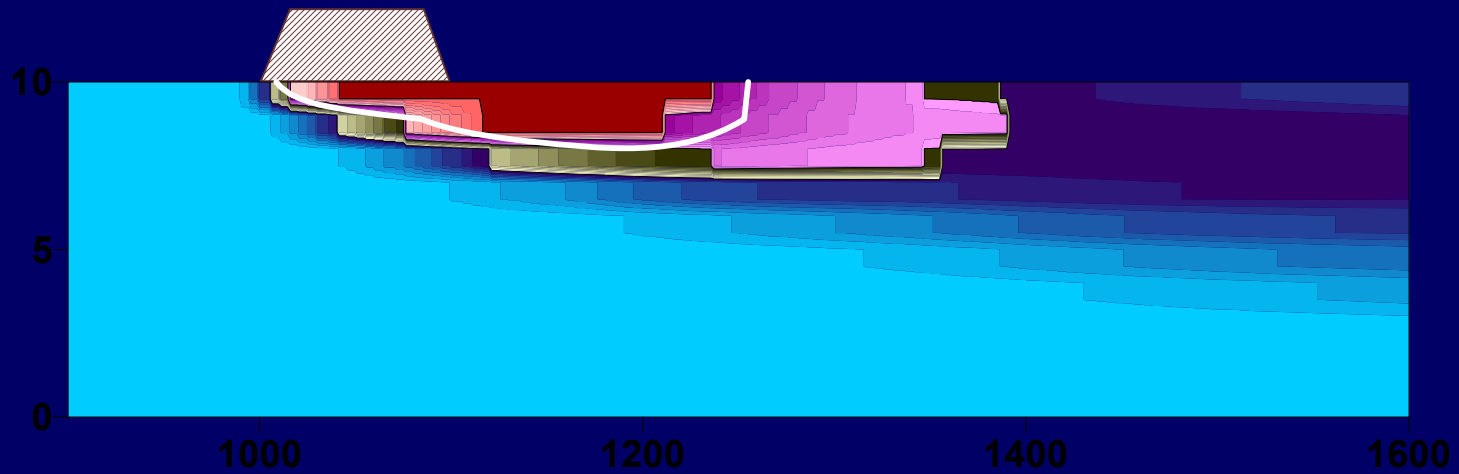
Permeable Cap (t = 17 y)



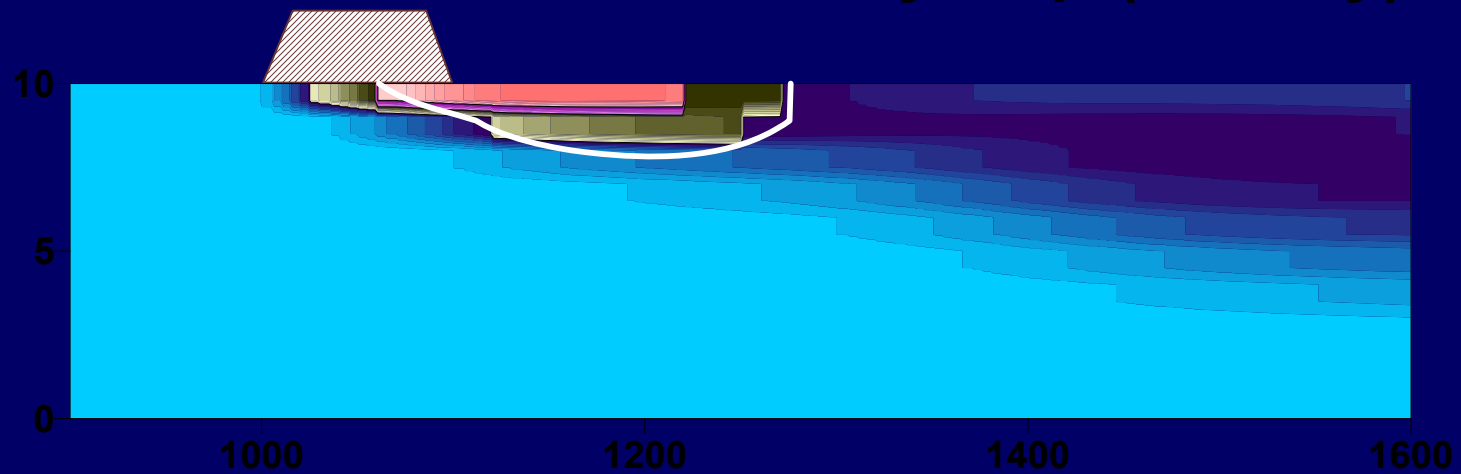
Clay Cap (t = 17 y)



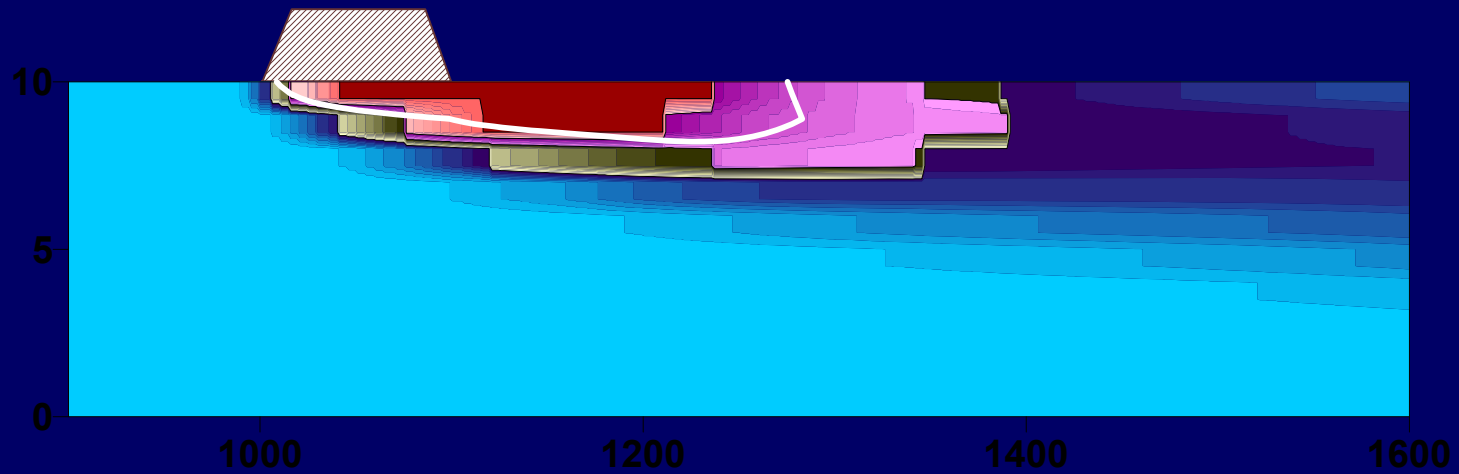
Permeable Cap (t = 17 y)



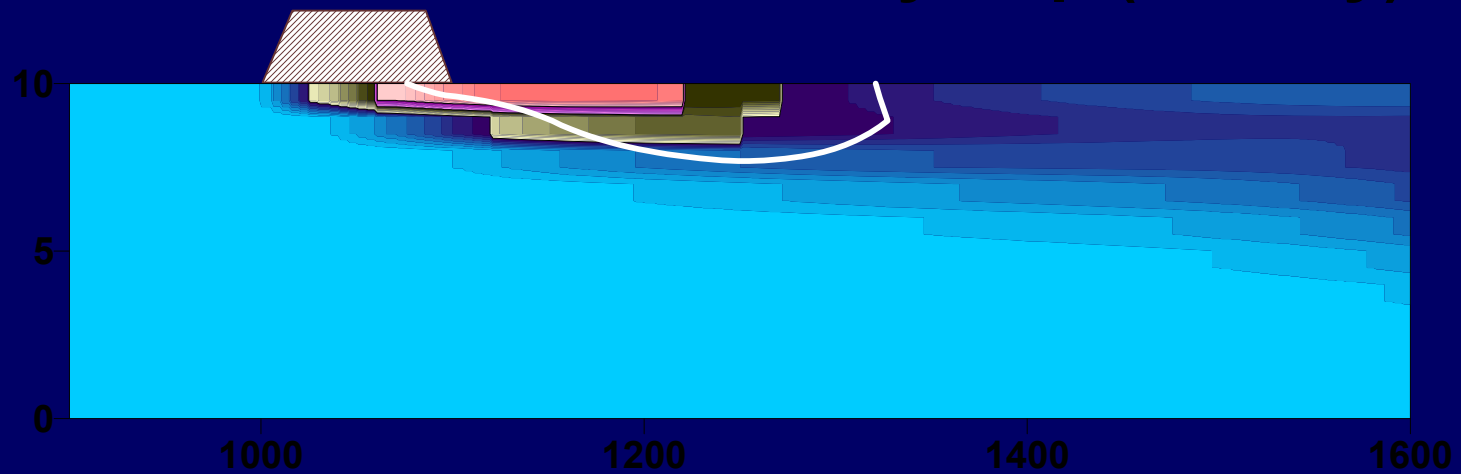
Clay Cap (t = 17 y)



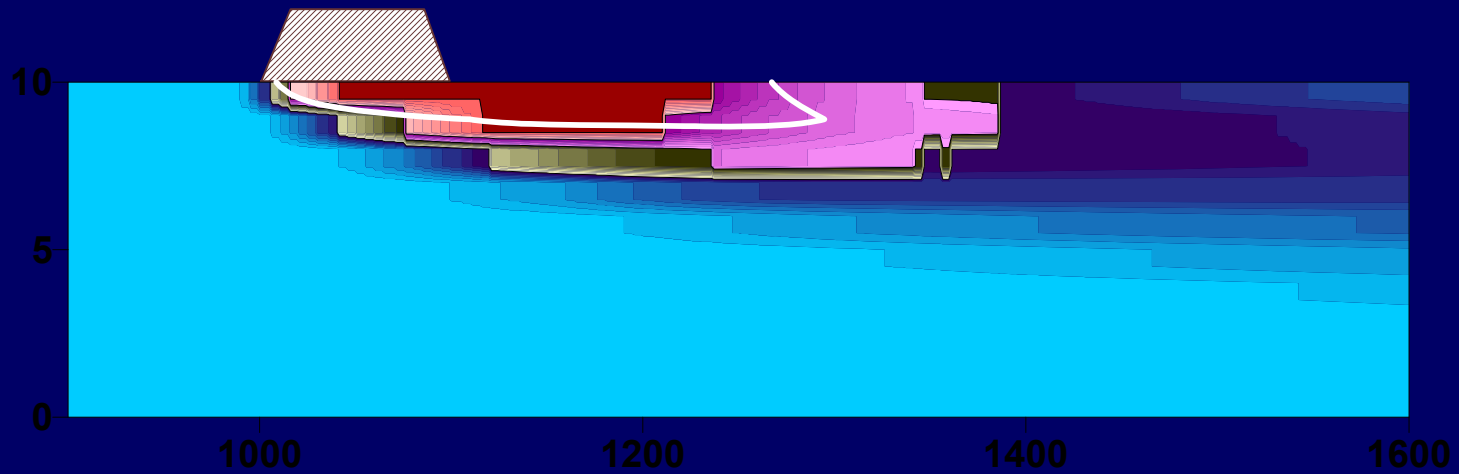
Permeable Cap (t = 18 y)



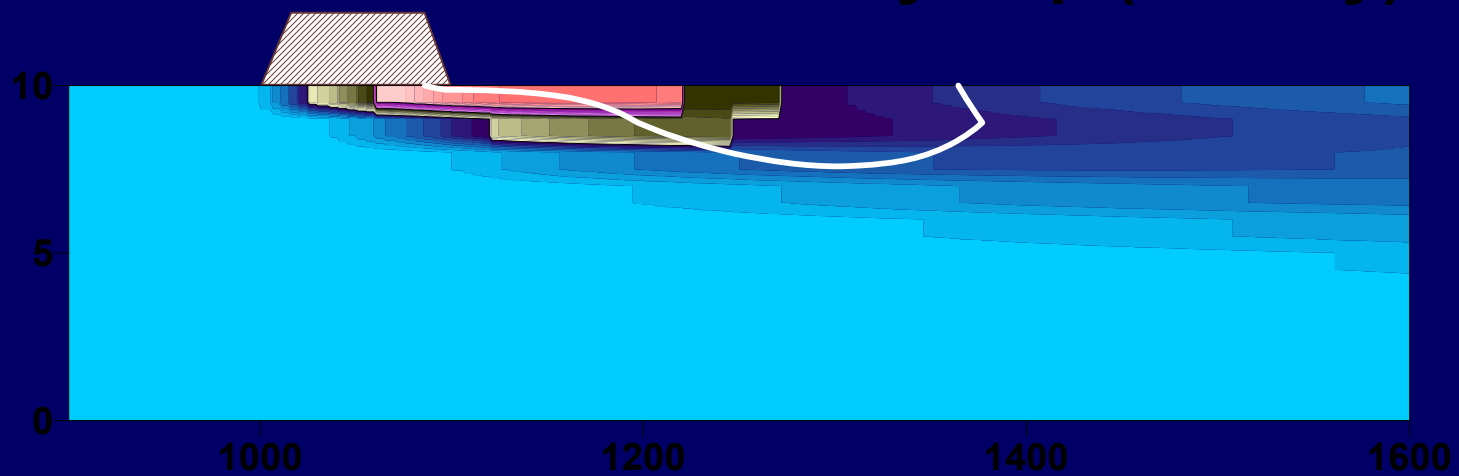
Clay Cap (t = 18 y)



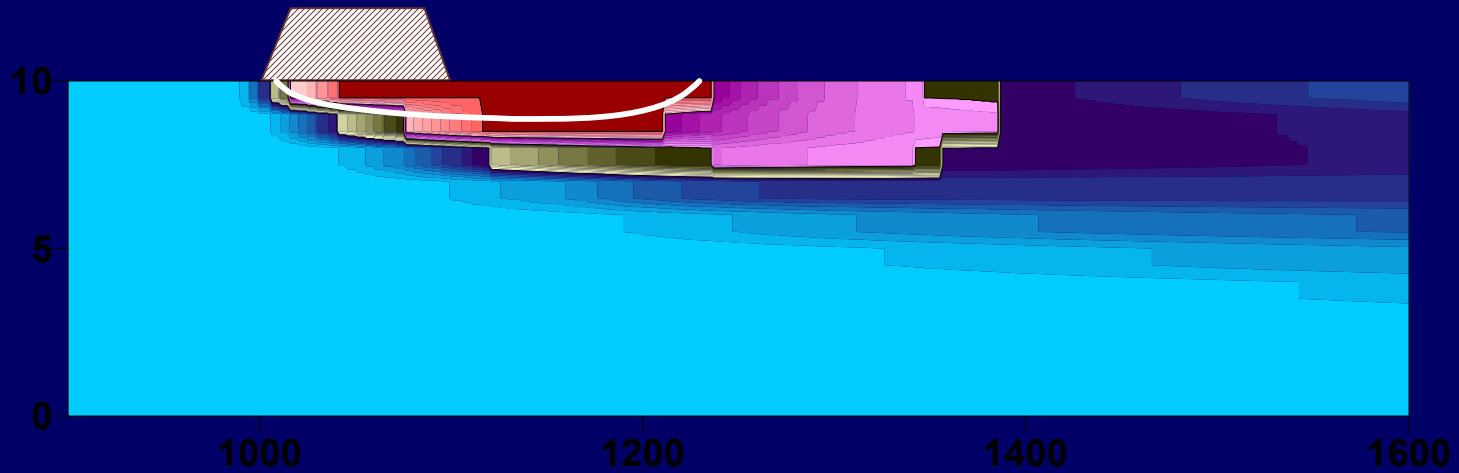
Permeable Cap (t = 19 y)



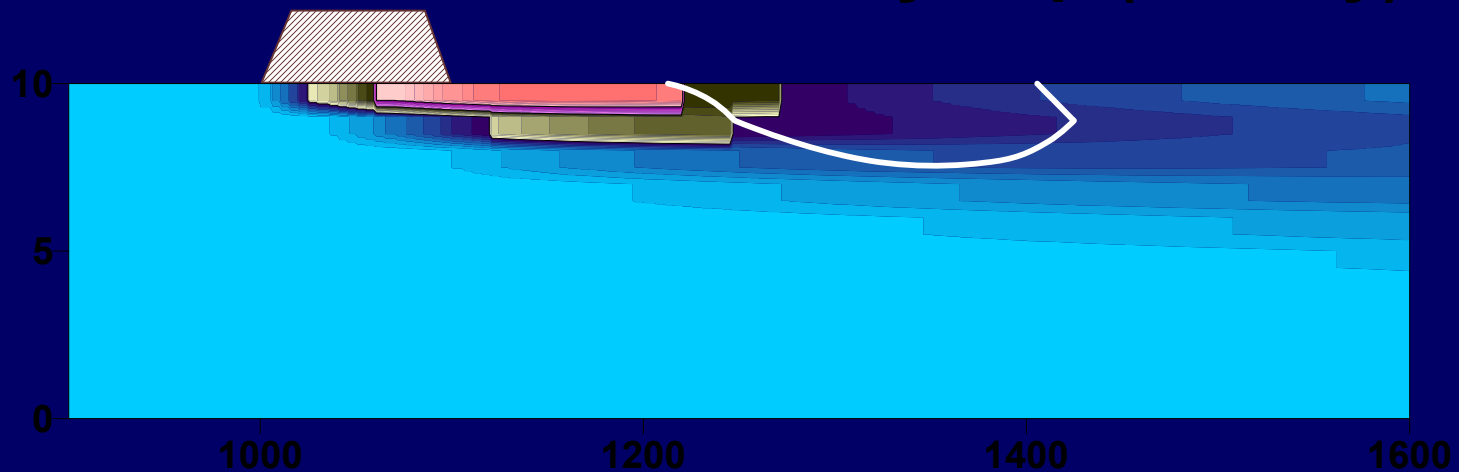
Clay Cap (t = 19 y)



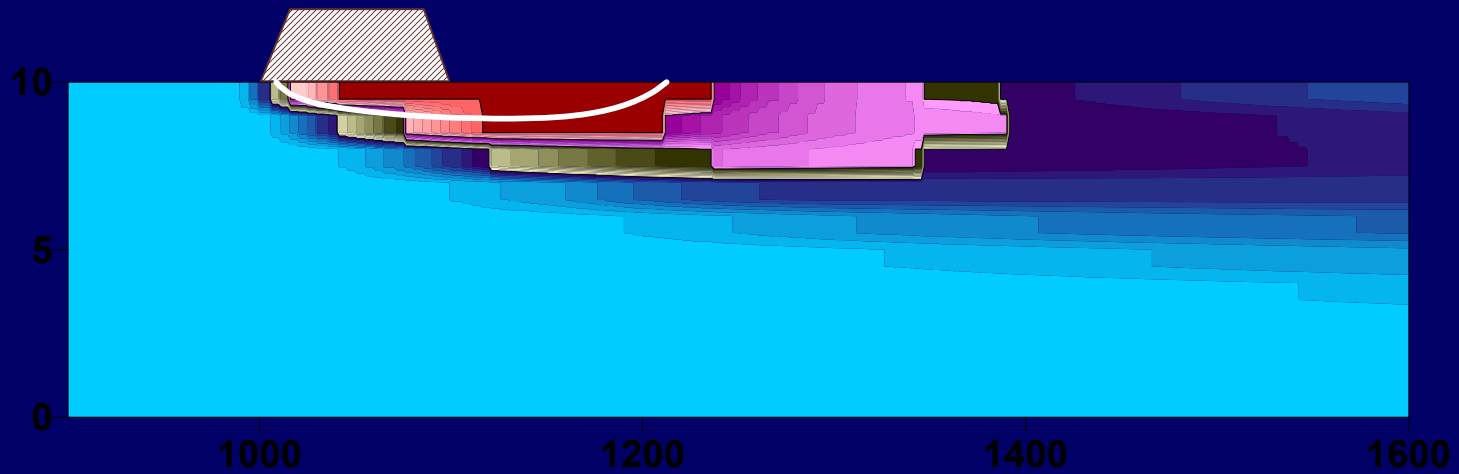
Permeable Cap (t = 20 y)



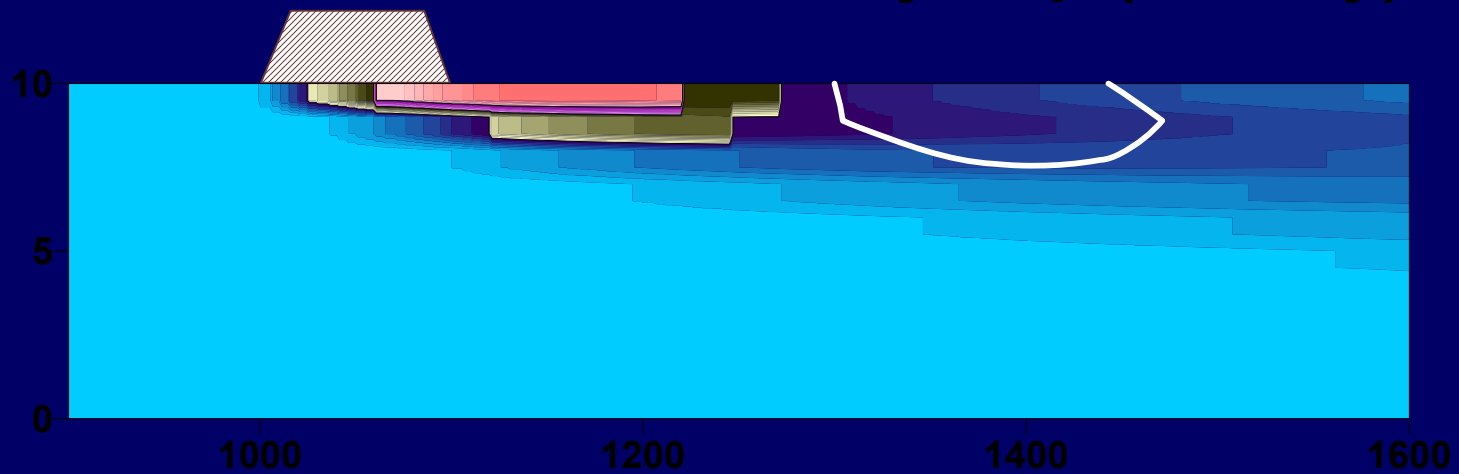
Clay Cap (t = 20 y)



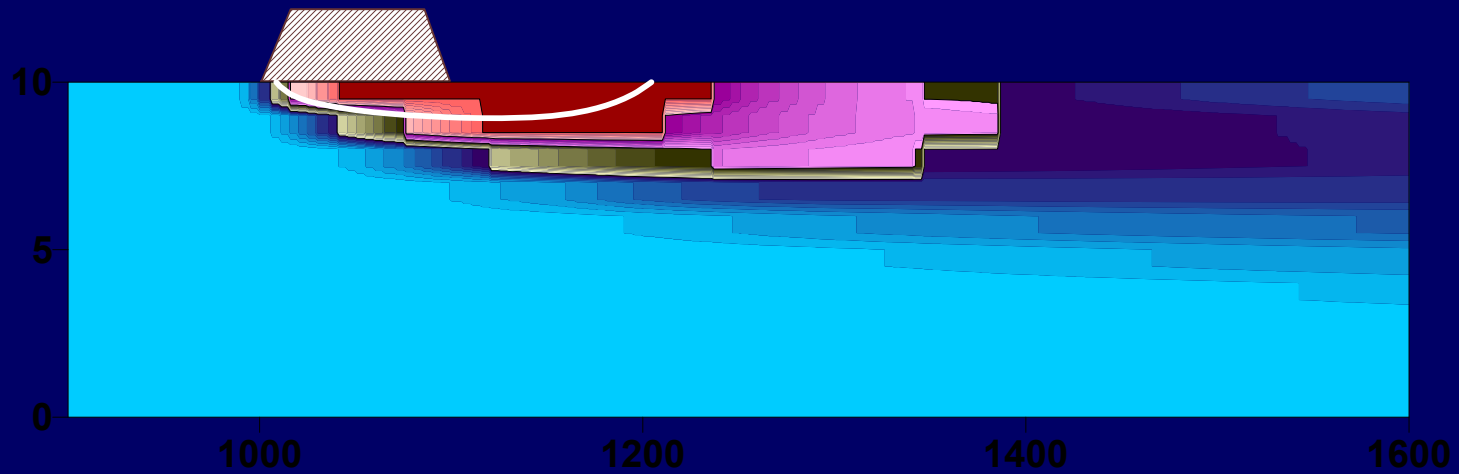
Permeable Cap (t = 21 y)



Clay Cap (t = 21 y)



Permeable Cap (t = 22 y)



Clay Cap (t = 22 y)

